

copper/nickel, copper/titanium, and copper/nickel/titanium, wherein the Cu-based alloy foil is produced by an electro-plating process;

a separator interposed between the positive and negative electrodes; and

an electrolyte into which the positive and negative electrodes and the separator are immersed.

2. (Twice Amended) The lithium secondary battery of claim 1 wherein an amount of nickel is 0.8 to 4 wt% of copper, an amount of titanium is 0.2 to 4 wt% of copper, an amount of magnesium is 0.05 to 0.6 wt% of copper, an amount of tin is 0.1 to 2.0 wt% of copper, an amount of zinc is 0.1 to 2.0 wt% of copper, an amount of manganese is 0.1 to 1.0 wt% of copper, an amount of iron or cobalt is 0.01 to 2.0 wt% of copper, and an amount of aluminum is 0.005 to 0.5 wt% of copper.

3. (Twice Amended) A method for making a lithium secondary battery comprising:

forming a positive electrode by coating a lithium metal oxide on a positive current collector;

forming a negative electrode by coating carbonaceous materials or  $\text{SnO}_2$  on a negative current collector where the negative current collector is made of a Cu-based alloy foil with a thickness of 20  $\mu\text{m}$  or less and the Cu-based alloy foil is prepared by adding at least one material selected from the group consisting of magnesium, tin, boron, chromium, manganese, cobalt, vanadium, zirconium, niobium, bismuth, and misch metal to a copper-based material selected from the group consisting of copper, copper/nickel, copper/titanium, and copper/nickel/titanium, wherein the Cu-based alloy foil is produced by an electro-plating process;

interposing a separator between the positive and negative electrodes; and

injecting an electrolyte to immerse the positive and negative electrodes and the separator.